

Diatomic Atoms
 H, N, O, F, Cl, Br, I

POLYATOMIC IONS CHART

Quizzes by
 1. Name the
 1. Compounds
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It would be a good idea to write down the some ionic compounds with polyatomic ions.

Polyatomic Ion	Chemical Symbol	Oxidation Number
ammonium	NH_4^+	1+
acetate	$C_2H_3O_2^-$	1-
bicarbonate	HCO_3^-	1-
bisulfate	HSO_4^-	1-
bisulfite	HSO_3^-	1-
bromate	BrO_3^-	1-
Bromite	BrO_2^-	1-
carbonate	CO_3^{2-}	2-
chlorate	ClO_3^-	1-
chlorite	ClO_2^-	1-
chromate	CrO_4^{2-}	2-
cyanide	CN^-	1-
dichromate	$Cr_2O_7^{2-}$	2-
ferrocyanide	$Fe(CN)_6^{4-}$	4-
cyanide	$Fe(CN)_6^{3-}$	3-
hydroxide	OH^-	1-
hypobromite	BrO^-	1-

On the left margin are transition metals with multiple charges.

This handout contains a list of polyatomic ions used for naming an forming some ionic compounds.

Sodium bicarbonate is compound containing the polyatomic ion bicarbonate.

Hydrogen Peroxide is another compound with a polyatomic ion.

Polyatomic Ion	Chemical Symbol	Oxidation Number
hypochlorite	ClO^-	1-
hypoiodate	IO^-	1-
iodate	IO_3^-	1-
Iodite	IO_2^-	1-
nitrate	NO_3^-	1-
nitrite	NO_2^-	1-
oxalate	$C_2O_4^{2-}$	2-
perbromate	$HBrO_3^-$	1-
perchlorate	ClO_4^-	1-
permanganate	MnO_4^-	1-
peroxide	O_2^{2-}	2-
phosphate	PO_4^{3-}	3-
sulfate	SO_4^{2-}	2-
sulfite	SO_3^{2-}	2-
thiosulfite	$S_2O_3^{2-}$	2-
hydrogen phosphate	HPO_4^{2-}	2-
dihydrogen phosphate	$H_2PO_4^-$	1-

Ag⁺ Zn²⁺ Cd²⁺ - only 3 w/ed multiple charge